# Condensation Reaction of 5, 6-Dihydro-6-methyl-6-piperonyl-2H-pyran-2, 4-dione, Ethyl Orthoformate and Substituted Anilines 

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#### Abstract

Piperonyl methyl ketone was obtained by oxidizing isosafrole with hydrogen peroxide and formic acid. Dianion of ethyl acetoacetate reacted with piperonyl methyl ketone and 5, 6-dihydro-6-methyl-6-piperonyl-2H-pyran-2, 4-dione was prepared, which reacted with substituted anilines in the presence of ethyl orthoformate to obtain 3-anilinomethylene-5, 6- dihydro6 -methyl-6-piperonyl-2H-pyran-2, 4-diones. Their structures were confirmed by ${ }^{1} \mathrm{HNMR}$ and elemental analysis.


Keywords: Piperonyl methyl ketone, isosafrole, dianion.

5,6-Dihydro-pyran-2,4-dione derivatives have interesting biological activities. Two American companies (Pharmacia \& Upjohn, Park-Davis Division of Warner-Lambert) had reported 5,6-dihydro-2H-pyran-2,4-dione derivatives have good inhibitory activity of HIV proteases ${ }^{1-3}$. Recently we have reported 3 -anilinomethylene-5, 6-dihydro-6-alkyl(aryl)-2H-pyran-2,4-dione derivatives have interesting fungicidal, tobacco virucidal activities ${ }^{4-5}$. The results led us to study the biological activities of different 5 , 6-dihydro-2H-pyran-2, 4-dione derivatives. We reacted 5, 6-dihydro-6-methyl-6-piperonyl-2H-pyran-2, 4-dione which was obtained by reacting dianion of ethyl acetoacetate with piperonyl methyl ketone with substituted anilines in the presence of ethyl orthoformate and new condensation products $\mathbf{3}$ were obtained.

In a general procedure, isosafrole was added dropwise to a solution of hydrogen peroxide and formic acid in the presence of 1,2 -dichloroethane at $40-45^{\circ} \mathrm{C}$. After the addition, the mixture was continued to stir for 2 h , the organic phase was separated from the solution and the solution was extracted two times with 1,2 -dichloroethane. The solvent was evaporated under the reduced pressure. The crude product was hydrolyzed with $15 \%$ diluted sulfuric acid for 3 h and piperonyl methyl ketone was obtained. Dianion of ethyl acetoacetate, which was obtained by reacted ethyl acetoacetate with NaH and ${ }^{\mathrm{n}} \mathrm{BuLi}$, reacted with piperonyl methyl ketone in the presence of absolute anhydrous THF for 6 h at $0^{\circ} \mathrm{C}$, and the slurry was poured into ice-cooled water. The solution was allowed to stir overnight and acidified with $5 \%$ diluted hydrochloride to obtain 5,6-dihydro-6-methyl-6-piperonyl-2H-pyran-2,4-dione. We reacted 5,6-

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a) $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}_{2} ; \quad$ b) $\mathrm{H}_{3}^{+} \mathrm{O}$.



dihydro-6-methyl-6-piperonyl-2H-pyran-2,4-dione(1) with substituted anilines(2) in the presence of ethyl orthoformate at $85-90^{\circ} \mathrm{C}$ and the products 3 were obtained. The products 3 were purified by silica gel column or recrystallization (Table 1). Their structures were confirmed by ${ }^{1} \mathrm{HNMR}$ (Table 2) spectra and elemental analysis (Table 1).

The products 3 that were confirmed by ${ }^{1} \mathrm{HNMR}$ spectra, are composed of a pair of isomers $\mathbf{Z}$ and $\mathbf{E}$, probably due to intramolecular hydrogen bonds. The chemical shifts of hydrogen on nitrogen atom and carbon-carbon double bond of $\mathbf{Z}$ lie in lower field because of the electron-withdrawing effect of oxygen atom.


Z-Form


E-Form

Condensation Reaction of 5, 6-Dihydro-6-methyl-6-piperonyl-2H- pyran-2, 223 4-dione, Ethyl Orthoformate and Substituted Anilines

Table1 Prepared data and elemental analysis data of 3-anilinomethylene-5,6-dihydro-6-methyl-6-piperonyl-2H-pyran-2,4-diones 3

| Compound | Xn | Yield (\%) | $\mathrm{m} . \mathrm{p}\left({ }^{\circ} \mathrm{C}\right)$ | Elemental analysis(\%, Cacl.) |  |  |
| :---: | :--- | :---: | :---: | :---: | :---: | :---: |
|  |  |  | C | H | N |  |
| $\mathbf{a}$ | $4-\mathrm{CH}_{3}$ | 74.9 | $149-151$ | $69.64(69.66)$ | $5.27(5.54)$ | $4.03(3.69)$ |
| $\mathbf{b}$ | $4-\mathrm{Cl}$ | 70.2 | $182-184$ | $62.96(63.08)$ | $4.52(4.51)$ | $3.35(3.50)$ |
| $\mathbf{c}$ | $4-\mathrm{Br}$ | 80.1 | $189-191$ | $56.67(56.76)$ | $4.16(4.05)$ | $3.10(3.15)$ |
| $\mathbf{d}$ | H | 66.4 | $173-175$ | $68.83(69.04)$ | $5.25(5.21)$ | $3.83(3.84)$ |
| $\mathbf{e}$ | $3,4-\mathrm{Cl}_{2}$ | 75.7 | $148-150$ | $57.90(58.06)$ | $3.94(3.92)$ | $3.15(3.23)$ |
| $\mathbf{f}$ | $2,5,6-\mathrm{Cl}_{3}$ | 79.6 | $198-200$ | $53.76(53.79)$ | $3.38(3.42)$ | $2.70(2.99)$ |
| $\mathbf{g}$ | $2-\mathrm{CH}_{3}$ | 64.3 | $117-119$ | $69.65(69.66)$ | $5.65(5.54)$ | $3.68(3.69)$ |
| $\mathbf{h}$ | $2-\mathrm{Br}^{2}$ | 82.7 | $157-159$ | $56.93(56.76)$ | $3.91(4.05)$ | $2.93(3.15)$ |
| $\mathbf{i}$ | $4-\mathrm{OCH}_{3}$ | 78.8 | $156-157$ | $66.72(66.84)$ | $5.19(5.32)$ | $3.76(3.54)$ |
| $\mathbf{j}$ | $2-\mathrm{OCH}_{3}$ | 68.7 | $168-170$ | $66.77(68.84)$ | $5.27(5.32)$ | $3.60(3.54)$ |
| $\mathbf{k}$ | $2-\mathrm{CH}_{3}-4-\mathrm{Cl}$ | 71.5 | $163-165$ | $63.66(63.85)$ | $5.12(4.84)$ | $3.38(3.39)$ |
| $\mathbf{l}$ | $2,6-\left(\mathrm{CH}_{3}\right)_{2}$ | 67.8 | Oil | $70.32(70.23)$ | $5.51(5.85)$ | $3.33(3.56)$ |
| $\mathbf{m}$ | $2-\mathrm{CH}_{3}-6-\mathrm{C}_{2}$ | 62.6 | Oil | $70.60(70.76)$ | $5.99(6.14)$ | $3.35(3.44)$ |
|  | $\mathrm{H}_{5}$ |  |  |  |  |  |

Table $2{ }^{1}$ HNMR data of 3-anilinomethylene-5,6-dihydro-6-methyl-6- piperonyl-2H-pyran-2, 4-diones 3

| compound | ${ }^{1} \mathrm{HNMR}, \delta, \mathrm{J}_{\mathrm{H}-\mathrm{H}} / \mathrm{Hz}$ |
| :---: | :---: |
| a | $1.43(\mathrm{~s}, 3 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}), 2.80(\mathrm{~m}, 4 \mathrm{H}), 5.89(\mathrm{~s}, 2 \mathrm{H}), 6.71(\mathrm{~m}, 3 \mathrm{H}), 7.19(\mathrm{~m}, 4 \mathrm{H}), 8.52(\mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}=12.0$ ), 11.45(d, J=14.0), 12.45(d, J=14.0) |
| b | $\begin{aligned} & 1.42(\mathrm{~s}, 3 \mathrm{H}), 2.74(\mathrm{~m}, 4 \mathrm{H}), 5.89(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 3 \mathrm{H}), 7.19(\mathrm{~m}, 4 \mathrm{H}), 8.51(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}=12.8) \text {, } \\ & 11.45(\mathrm{~d}, \mathrm{~J}=14.0), 12.45(\mathrm{~d}, \mathrm{~J}=14.0) \end{aligned}$ |
| c | $\begin{aligned} & 1.42(\mathrm{~s}, 3 \mathrm{H}), 2.88(\mathrm{~m}, 4 \mathrm{H}), 5.90(\mathrm{~s}, 2 \mathrm{H}), 6.75(\mathrm{~m}, 3 \mathrm{H}), 7.20(\mathrm{~m}, 4 \mathrm{H}), 8.49(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}=11.4) \text {, } \\ & 11.44(\mathrm{~d}, \mathrm{~J}=14.6), 12.43(\mathrm{~d}, \mathrm{~J}=14.6) \end{aligned}$ |
| d | $\begin{aligned} & 1.43(\mathrm{~s}, 3 \mathrm{H}), 2.70(\mathrm{~m}, 4 \mathrm{H}), 5.90(\mathrm{~s}, 2 \mathrm{H}), 6.71(\mathrm{~m}, 3 \mathrm{H}), 7.24(\mathrm{~m}, 5 \mathrm{H}), 8.55(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}=11.4) \text {, } \\ & 11.49(\mathrm{~d}, \mathrm{~J}=12.6), 12.45(\mathrm{~d}, \mathrm{~J}=12.6) \end{aligned}$ |
| e | $\begin{aligned} & 1.46(\mathrm{~s}, 3 \mathrm{H}), 2.91(\mathrm{~m}, 4 \mathrm{H}), 5.93(\mathrm{~s}, 2 \mathrm{H}), 6.73(\mathrm{~m}, 3 \mathrm{H}), 7.37(\mathrm{~m}, 3 \mathrm{H}), 8.42(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}=10.8) \text {, } \\ & 11.41(\mathrm{~d}, \mathrm{~J}=13.6), 12.39(\mathrm{~d}, \mathrm{~J}=13.6) \end{aligned}$ |
| f | $\begin{aligned} & 1.44(\mathrm{~s}, 3 \mathrm{H}), 2.70(\mathrm{~m}, 4 \mathrm{H}), 5.88(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 3 \mathrm{H}), 7.53(\mathrm{~m}, 3 \mathrm{H}), 8.42(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}=10.4) \text {, } \\ & 11.82(\mathrm{~d}, \mathrm{~J}=12.6), 12.68(\mathrm{~d}, \mathrm{~J}=12.6) \end{aligned}$ |
| g | $\begin{aligned} & 1.43(\mathrm{~s}, 3 \mathrm{H}), 2.39(\mathrm{~s}, 3 \mathrm{H}), 2.78(\mathrm{~m}, 4 \mathrm{H}), 5.86(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 3 \mathrm{H}), 7.23(\mathrm{~m}, 4 \mathrm{H}), 8.56(\mathrm{dd}, 1 \mathrm{H}, \\ & \mathrm{J}=10.8), 11.71(\mathrm{~d}, \mathrm{~J}=13.6), 12.45(\mathrm{~d}, \mathrm{~J}=13.6) \end{aligned}$ |
| h | $\begin{aligned} & 1.43(\mathrm{~s}, 3 \mathrm{H}), 2.88(\mathrm{~m}, 4 \mathrm{H}), 5.87(\mathrm{~s}, 2 \mathrm{H}), 6.72(\mathrm{~m}, 3 \mathrm{H}), 7.35(\mathrm{~m}, 4 \mathrm{H}), 8.52(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}=10.8) \text {, } \\ & 11.88(\mathrm{~d}, \mathrm{~J}=14.6), 12.76(\mathrm{~d}, \mathrm{~J}=14.6) \end{aligned}$ |
| i | $\begin{aligned} & 1.42(\mathrm{~s}, 3 \mathrm{H}), 2.87(\mathrm{~m}, 4 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 5.88(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 3 \mathrm{H}), 7.10(\mathrm{~m}, 4 \mathrm{H}), 8.45(\mathrm{dd}, 1 \mathrm{H}, \\ & \mathrm{J}=12.6), 11.47(\mathrm{~d}, \mathrm{~J}=14.6), 12.48(\mathrm{~d}, \mathrm{~J}=14.6) \end{aligned}$ |
| j | $\begin{aligned} & 1.42(\mathrm{~s}, 3 \mathrm{H}), 2.77(\mathrm{~m}, 4 \mathrm{H}), 3.95(\mathrm{~s}, 3 \mathrm{H}), 5.88(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 3 \mathrm{H}), 7.10(\mathrm{~m}, 4 \mathrm{H}), 8.59(\mathrm{dd}, 1 \mathrm{H}, \\ & \mathrm{J}=12.6), 11.79(\mathrm{~d}, \mathrm{~J}=14.6), 12.66(\mathrm{~d}, \mathrm{~J}=14.6) \end{aligned}$ |
| k | $\begin{aligned} & 1.43(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 3 \mathrm{H}), 2.88(\mathrm{~m}, 4 \mathrm{H}), 5.85(\mathrm{~s}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 3 \mathrm{H}), 7.22(\mathrm{~m}, 4 \mathrm{H}), 8.48(\mathrm{dd}, 1 \mathrm{H}, \\ & \mathrm{J}=12.0), 11.66(\mathrm{~d}, \mathrm{~J}=14.0), 12.63(\mathrm{~d}, \mathrm{~J}=14.0) \end{aligned}$ |
| 1 | $\begin{aligned} & 1.43(\mathrm{~s}, 3 \mathrm{H}), 2.41(\mathrm{~s}, 6 \mathrm{H}), 2.76(\mathrm{~m}, 4 \mathrm{H}), 5.99(\mathrm{~s}, 2 \mathrm{H}), 6.77(\mathrm{~m}, 3 \mathrm{H}), 7.20(\mathrm{~m}, 3 \mathrm{H}), 8.10(\mathrm{dd}, 1 \mathrm{H} \text {, } \\ & \mathrm{J}=11.6), 11.01(\mathrm{~d}, \mathrm{~J}=14.0), 11.98(\mathrm{~d}, \mathrm{~J}=14.0) \end{aligned}$ |
| m | $\begin{aligned} & 1.13(\mathrm{~m}, 5 \mathrm{H}), 1.31(\mathrm{~s}, 3 \mathrm{H}), 2.54(\mathrm{~m}, 4 \mathrm{H}), 5.96(\mathrm{~s}, 2 \mathrm{H}), 6.76(\mathrm{~m}, 3 \mathrm{H}), 7.18(\mathrm{~m}, 3 \mathrm{H}), 7.85(\mathrm{dd}, 1 \mathrm{H}, \mathrm{~J}= \\ & 11.6), 10.91(\mathrm{~d}, 14.6), 11.82(\mathrm{~d}, \mathrm{~J}=14.6) \end{aligned}$ |

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